

Physicochemical Properties of *Silybum Marianum* Seed Oil in Two Different Regions of Iran

Z. Nemati*, E. Talebi, I. Nasrollahi, M. Khosravinezhad

Abstract— In this experiment, *Silybum marianum* was collected from Kazeron and Lorestan in Iran. Hexane Soxhlet method were used and the oil was analyzed by gas chromatography. The fatty acid composition included C16:0, C18:0, C18:1, C18:2 were estimated. The oil refractive index was performed with Refractometr. Peroxide was measured with thiosulfate and peroxide value of the oil sample was 0.68 and 0.57 in Kazeron and Lorestan, respectively.

Index Terms— Methyl esters, Peroxide, *Silybum marianum*, Soxhlet, Thiosulfate.

I. INTRODUCTION

One of the important medicinal plant is milk thistle which contain nutritional and medicinal applications. Edible fats, which are suitable for certain food service and bakery purposes, need to be solid or semisolid at room temperature. Those commonly used are animal fats such as tallow, vegetable fats and blends containing palm oil, coconut oil, and partially hydrogenated soybean oil. It is shown that the oil extracted from leaves and seeds of *Silybum marianum* plant can prevent liver diseases; and promote antioxidant and anti-cancer effects. The health disadvantage cause in high content of lauric, myristic, palmitic, or trans fatty acids which has been shown to raise the plasma low-density-lipoprotein (LDL) concentrations in people [1, 2]. In the last years, new analytical methods have been used to evaluate the processing and storage conditions of oil [3,4,5,6,7]. The quality of edible oils with regard to freshness, storability and toxicity can be evaluated by the determination of metals. Trace elements such as Fe, Cu, Ca, Mg, Co, Ni and Mn are known to increase the rate of oil oxidation compare to other elements such as Cr, Cd, and Pb which they are very important on account of toxicity and metabolic role. Thus, the development of rapid and accurate analytical methods for determination of trace elements in oil has been a challenge for quality control and food analysis [8]. Milk thistle or *Silybum marianum* is well known as medicinal plant that is native to Mediterranean region of Europe and widely dispersed to many countries throughout the world [9,10]. *Silybum Marianum* is a medicinal plant which has been used for centuries as herbal medicine treatment of some liver diseases and it is a biennial in nature or annual in cultivation medicinal plant that has been widely used in the European traditional medicine [11]. It is widely prescribed by herbalists that we still didn't have information on its parallel effects. This plant is native in Mediterranean and grows throughout Europe and North America as well as grows in India, China, South

America, Africa and Australia [12]. *Silymarin*, as a flavonolignan complex that contains silibinin, was isolated from this plant in the 17th century and has been clinically used to treat various liver ailments for more than three decades [13].

Seed oil chemical composition of wild growing *Silybum marianum* was studied for the first time in Bulgaria. In this research, physicochemical properties and fatty acid composition of extracted oil from milk thistle seeds were studied to identify the composition of fatty acids and measurement of the refractive index for determination of peroxide oil which collected from two different regions of Iran namely Lorestan (Khoramabad) (west of Iran) and Kazeron (south of Iran).

II. COLLECTION OF SAMPLES

Milk thistle seeds were collected in Kazeron (with 860 m height from sea) and Lorestan (with 500 m height from sea) regions (Figure 1) in July 2014. The plants were identified by Medicinal Plant Research Center, Shiraz University. The seeds were dried at room temperature (25°C). For extraction of oil from the seeds, thistle and hexane soxhlet apparatus was used.



Figure 1 Geographical location of Kazeron and Lorestan

III. REAGENTS AND CHEMICALS

In this experiment, the chemicals were used including hexane, acetic acid, chloroform, potassium iodine, thiosulfate (Merck, Germany).

Oil extraction

The conventional oil extraction, a pilot-scale soxhlet extraction unit was operated for 9 h using AW406, which is a commercial organic solvent applied to the extraction of crude edible oils. Based on the manufactory data, the *n*-hexane in

Z. Nemati*, Shiraz branch, Islamic Azad University, Shiraz, Fars, Iran
E. Talebi, Darab Branch, Islamic Azad University, Darab, Fars, Iran
I. Nasrollahi, 3 Young Researchers and Elite Club, Darab Branch, Islamic Azad University, Darab, Fars, Iran
M. Khosravinezhad, Shiraz branch, Islamic Azad University, Shiraz, Fars, Iran

solvent was 30% (with a total hexane content of 70%) [14] (Figure 2).

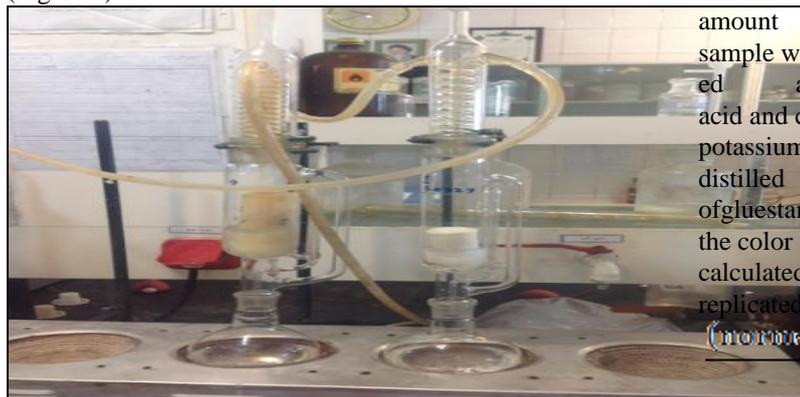


Figure 2 *Silybum marianum* seed oil extraction by Soxhlet and hexane

Analysis of fatty acids

The fatty acid composition was determined by conversion of oil to fatty acid methyl esters prepared (mixing of 950 μ L of n-hexane and 50 mg of oil followed by adding 50 μ L of sodium methoxide) [15]. Fatty acids were transformed to their methyl esters (FAME), following the method of He and Xia [16] and determined by using a gas chromatography-trace series (PEG20 M) equipped with a flame ionization detector. The FAME sample (1.5 μ L) was injected and GC separation was carried out on a capillary column. The carrier gas was nitrogen and the column flow rate was 0.8 mL/min.

The oven temperature was held initially at 180 $^{\circ}$ C for 1 min, increased from 180 to 240 $^{\circ}$ C at a rate of 3 $^{\circ}$ C/min and then maintained at 240 $^{\circ}$ C for 10 min. The temperatures of injection port and detector were 250 and 260 $^{\circ}$ C, respectively. FAME were positively identified by matching their retention time data and mass spectra with those of the standards. The fatty acid composition was calculated from the total identified fatty acid area [17].

Measuring the refractive index of oil

A drop of oil in a refractometer using the refractive index of oil compared to the refractive index of water is calculated at 40 $^{\circ}$ C (Figure 3).



Figure 3 Measurement of refractive index of oil

Determination of peroxide oil

The peroxide value estimated based on standard tests of food (AOAC) [18]. In this method, the amount of 5 g of sample was prepared in a 250 mL flask, weighed and 30 mL of solvent (mixture of acetic acid and chloroform) added, then about 5 mL of potassium iodide added and mixed for 1 minute. 30 mL of distilled water and solution was added, then a few drops of glucose starch applied using soluble thiosulfate 0.02%. When the color of a transparent cleared, peroxide values calculated by following formula [18]. Experiments were replicated three times for each sample (Figure 4).

$$\text{Peroxide Value} = \frac{\text{Normality} \times \text{Dose of titration volume} \times 1000}{\text{Sample volume}}$$

Sample volumes



Figure 4 Measurement of oil peroxide

IV. RESULTS AND DISCUSSION

Lipids are considered one of the most elemental nutrients for humans and animals. Lipid metabolism generates many bioactive lipid molecules, which are fundamental mediators of multiple signaling pathways and they are also indispensable compounds of cell membranes. The total oil content of the seeds was 25.32% and the fatty acids of *Silybum marianum* seed oil were identified by GC. The *Silybum marianum* seeds, which were collected from two locations, namely Kazeran and Lorestan, exhibited C16:0, C18:0, C18:1, C18:2 (Table 1) with an average of 25.32% oil yield. This finding is according with the previous reports on *Silybum marianum* seeds that reported variations in oil yield and content which may be due to the differences in variety of plant, cultivate on climate, ripening stage, the harvesting time of the seeds, location and the extraction method [19, 20]. *Silybum marianum* seed oil also contained linoleic (54.71%) and oleic (30.42%) acids as the principal fatty acids [21, 22]. The linoleic acid was the main fatty acid followed by oleic 52.1% and 32.14%, respectively. For the oil extracted from an unspecified variety of *Silybum marianum* seed oil also reported that linoleic acid (52.78%) was a principle fatty acid [23, 24].

Table 1 Fatty acid composition of *Silybum marianum* oil (%)

Fatty acid composition	acid	Kazer on	Lorest an
C1 6:0	Palmitic acid	7.99	9.26
C1 6:1	Palmitoleic acid	-	-
C1 7:1	Margaric acid	-	-
C1 8:0	Stearic acid	5.607	5.01
C1 8:1	Oleic acid	28.54	30.42
C1 8:2	Linoleic acid	54.71	52.78
C1 8:3	α -Linolenic acid	3.13	2.51
C2 0:0	Arachidic acid	-	-
C2 2:0	dodecanoic acid	-	-

The oil refractive index of *Silybum marianum* was estimated with Refractometer. The results were shown in below (Table 2). The fatty acid double bond is less compared to the number of lower refractive index as well as the refractive index is higher than the melting point with lower saturation. The physicochemical properties of oil that harvested in Kazeran and Lorestan were as follows: refractive index 1.4651 and 1.4656 (Table 2) and peroxides oil, 0.68 and 0.57 (Table 3) respectively.

Table 2. The refractive index of *Silybum marianum* seed oil

Location of collecting sample	Refractive index
Kazeran	1.4651
Lorestan	1.4656

The concentrations of peroxides and hydroperoxides were measured at the start of experiment. The peroxide value was estimated by taking expiration date of the product, normal oil index [25, 26]. The measurement of oil peroxides are shown in Table 3.

Table 3. Measuring the peroxide sample

Sample	peroxide (meq/kg)
Kazeran	0.68
Lorestan	0.57

The present study established similar compounds in *Silybum marianum* seed oil, some of them have valuable applications as a food and pharmaceuticals. We found that the main components in fatty acids were oleic and linoleic acids.

V. ACKNOWLEDGEMENTS

The authors wish to express sincere thanks to the Faculty and Chairman (Head of the Department), Department of Animal Science, Darab branch, Islamic Azad University, Darab, Fars, IRAN and Herbal Medicine Research Center, Shiraz University, Iran for extending the laboratory facilities to carry out the research work.

ACKNOWLEDGMENT

The preferred spelling of the word "acknowledgment" in American English is without an "e" after the "g." Use the singular heading even if you have many acknowledgments. Avoid expressions such as "One of us (S.B.A.) would like to thank" Instead, write "F. A. Author thanks" **Sponsor and financial support acknowledgments are placed in the unnumbered footnote on the first page.**

REFERENCES

- [1] Hegsted DM, McOandy RB, Myers SM, Stare FJ. Quantitative effects of dietary fat on blood cholesterol in man. *Am J Clin Nutr.*, 1965;17: 281-295.
- [2] Nestel PJ, Noakes M, Belling GB, et al. Plasma lipoprotein lipid and Lp(a) changes with substitution of elaidic acid for oleic acid in the diet. *J Lipid Res* 1992;33:1029-36.
- [3] Lee S, Min DB. Effects, quenching, mechanisms, and kinetics of carotenoids in chlorophyll-sensitized photooxidation of soybean oil. *J Agric Food Chem.*, 1990; 38:1630-1634.
- [4] Ramamurthi S, Low HN. Effect of possible chlorophyll breakdown products on canola oil stability. *J Agric Food Chem.*, 1995; 43:1479-1483.
- [5] Oomah BD, Liang J, Godfrey D, Mazza G. Microwave heating of rapeseed: effect on oil quality. *J Agric Food Chem.*, 1998; 46:4017-4021.
- [6] Gennaro L, Bocca AP, Modesti D, Masella R, Coni E. Effect of biophenols on olive oil stability evaluated by thermo gravimetric analysis. *J Agric Food Chem.*, 1998; 46:4465-4469.
- [7] Vianni R, Braz-Filho R. Ácidos graxos naturais: importância e ocorrência em alimentos. *Quím. Nova*, 1996; 19:400-407.
- [8] Anthemidis A.N., V. Arvanitidis, J.A. Stratis. On-line emulsion formation and multi-element analysis of edible oils by inductively coupled plasma atomic emission spectrometry. *Analytica Chimica Acta*, 2005; 537: 271-278.
- [9] Frascini F, Demartini G, and Esposti D. Pharmacology of silymarin. *Clin. Drug Invest.*, 2002; 22(1): 51-65.
- [10] Kurkin VA. Saint-Mary thistle: a source of medicinals. *Pharmaceut. Chem. J.*, 2003; 37(4): 189-202.

- [11] Rasul A, Akhtar N, Khan BA, TariMahmood, Zaman S, Atif A, Haji M, Khan S, Parveen S, Assessment of anti erythmic and skin whitening effects of milk thistle extract. African Journal of Pharmacy and Pharmacology, 2011; 5(20), 2306-2309.
- [12] Dixit N, Baboota S, Kohli K, Ahmed S, Javed A, Silymarin, A review of pharmacological aspects and bioavailability enhancement approaches. Indian J. Pharmacol., 2007; 39(4),172-179.
- [13] Singh RP, Agarwal R, Cosmeceuticals and silibinin. Clinics in Dermatology, 2009; 27, 479–484.
- [14] EhsanJenab ,KaramatollahRezaei ,Zahra Emam-Djomeh, Canola oil extracted by supercritical carbon dioxide and a commercial organic solvent . Eur. J. Lipid Sci. Technol.,2006;108. 488–492.
- [15] Cocks LV, Van Rede C, (1966). Laboratory Handbook for Oil and Fats Analysts. Pp. 305–314. London, UK: Academic Press
- [16] He ZY, Xia WS, Nutritional composition of the kernels from Canarium album L. Food Chem., 2007; 102:808–811.
- [17] ZhengLianhe, Huang Xing, Wang Li, Chen Zhengxing, Physicochemical Properties, Chemical Composition and Antioxidant Activity of Dalbergiaodorifera T. Chen Seed Oil . J. Am. Oil Chem. Soc., 2012; 89: 883–890
- [18] Horwitz W, Senze A, Reynolds H, Park DL. (1975). Official methods of analysis of the association of analytical chemists. Washington: Associat Official Analytic Chemist.
- [19] Abdulkarim SM, Long K, Lai OM, Muhammad SKS, Ghazali HM. Some physico-chemical properties of Moringaoleifera seed oil extracted using solvent and aqueous enzymatic methods. Food Chem., 2005; 93:253–263.
- [20] Gecgel U, Demirci M, Esendal E, Tasan M. Fatty acid composition of the oil from developing seeds of different varieties of safflower (Carthamustinctorius L.). J. Am. Oil Chem. Soc., 2007; 84:47–54.
- [21] De Mello MLS, Bora PS,Narain N. Fatty and amino acids composition of melon (Cucumismelovar. saccharinus) seeds. J. Food Comp. Anal., 2001; 14: 69–74.
- [22] Hemavatahy J. Lipid composition of melon (Cucumismelo) kernel. J. Food Comp. Anal., 1992; 5: 90–95.
- [23] EL-Magoli SB, Morad MM, EL-Fara AA, Evaluation of some Egyptian sweet melon seed oils. Fette. Seifen. Anstrichmittel, 1979; 8: 201–203.
- [24] EL-Adawy TA,Taha KM, Characteristics and compositions of different seed oils and flours. Food Chem., 2001; 74: 47–54.
- [25] Parvaneh V, Quality control and chemical analyses of food: edible oils and fats. Tehran: Tehran University Pub. 2005.Persian.
- [26] Matthaus B, Utilization of high oleic rapeseed oil for deep fat frying of French fries compared to other commonly used edible oils. Europ J Lipid Sci Tech., 2006; 108(3): 200-211.